

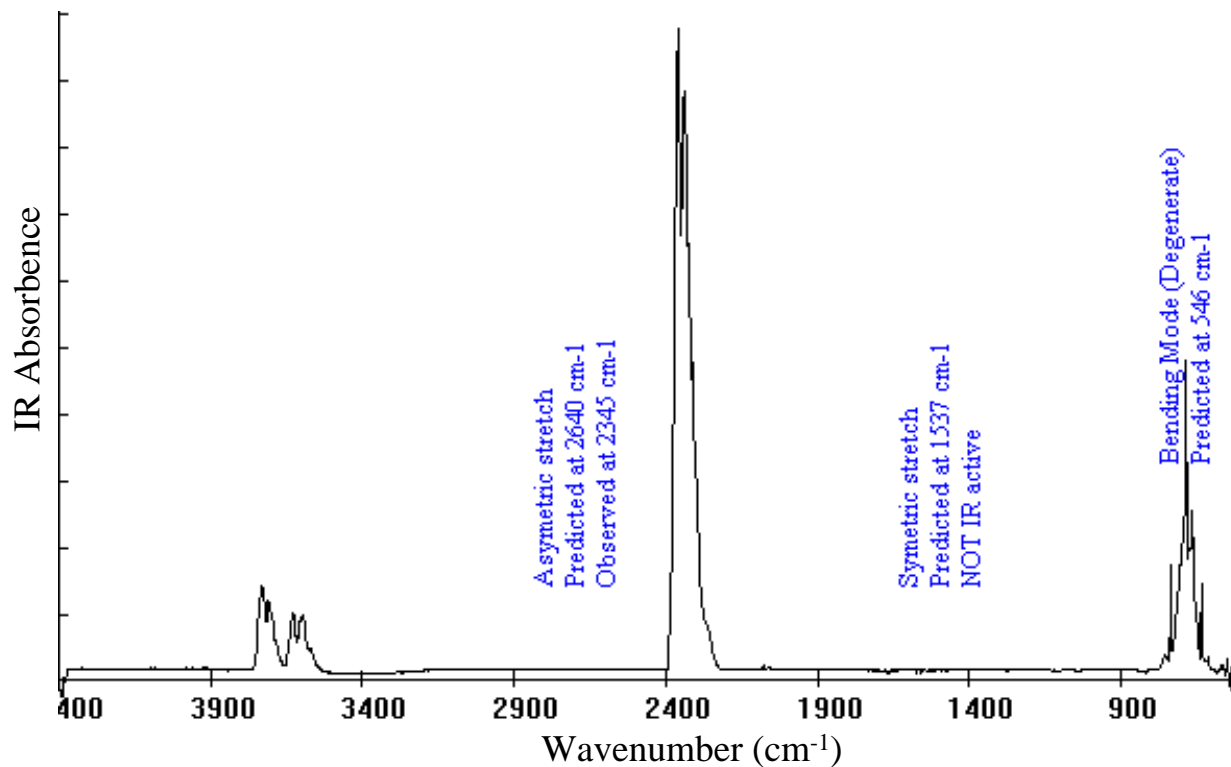
What can we learn from IR spectroscopy?

- Atoms vibrate with frequencies in the IR range
- **Chemical Analysis:**
 - Match spectra to known databases
 - Identifying an unknown compound, Forensics, etc.
 - Monitor chemical reactions *in-situ*
- **Structural ideas:**
 - Can determine what chemical groups are in a specific compound
- **Electronic Information:**
 - Measure optical conductivity
 - Determine if Metal, Insulator, Superconductor, Semiconductor
 - Band Gaps, Drude model

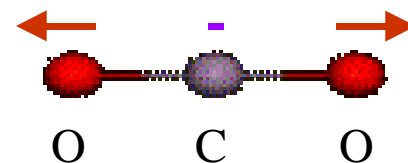


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An Example: CO₂



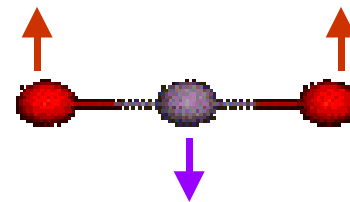
A Dipole Moment = charge imbalance in the molecule



Symmetric Stretch
(Dipole moment = 0 so **not** IR active)



Asymmetric Stretch
(Has dipole moment so IR active)



Bending Mode
(Has dipole moment so IR active)